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answers.pdf. Topic 1:

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Textbook pa, 42 - 47.

1. What are the three

main states of

matter? a - Solids b.

Liguide c-Gases.

2. Match the following

characteristics with

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the correct state of matter, there may be more than one answer for each characteristic. C No fixed shape & volume  
A. Solids.

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Key gases. The particles in a gas are considered to be small, hard spheres with an insignificant volume. Between particles in a gas there is empty space. No attractive or repulsive forces exist between the particles.

3 PPT – Chapter 13

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Answer Key

No attractive or  
repulsive forces exist  
between the particles.

3 Chapter 13 States  
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Theory as Applied to  
Gases Fundamental  
assumptions about

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gases. The particles in a gas are considered to be small, hard spheres with an insignificant volume. Chapter 13 States Of Matter Worksheet Look at the text on page 315 for the answer.

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relationship among  
the solid, liquid, and  
vapor states (or  
phases) of a

substance in a sealed  
container are best

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Answers 1. Are these  
examples of physical  
or Chemical Change?

a. burning paper. A  
Change from liquid to



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gas. 13. An uncovered pot of soup is simmering on a stove, and there are water droplets on the wall above the back of the stove.

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system at constant vapor pressure, a dynamic equilibrium exists between the vapor and the liquid. The system is in equilibrium because the rate of evaporation of liquid equals the rate of condensation of vapor.

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