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**Of Chemical Bonds - What Are**

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**Bonds And Ionic Bonds - What**

**Are Ions GCSE Chemistry -**

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*Covalent bonding in methane,*

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ammonia \u0026amp; water |  
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this set (68) When sharing

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of electrons occurs the attachment between atoms is called covalent bond in a covalent bond, the dissociation energy is released in the process of exothermic reaction When 2 or more atoms bond by means

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Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure

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8.4, each covalently bonded atom equally attracts the pair of shared electrons.

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Covalent Bonding 219 You can draw the electron dot structure for ammonia ( $\text{NH}_3$ ), a suffocating gas, in a similar way. The ammonia molecule has one unshared pair of electrons. Hydrogen atoms 3H. or Nitrogen atom

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Ammonia molecule N...H  
H.H.... H H H.N HH 1s 1s 1s  
2s 2p N H 1s Ammonia  
molecule Methane (CH<sub>4</sub>)  
contains four single  
covalent bonds. The carbon  
atom

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8.2 The Nature of Covalent Bonding 8

242 Chapter 8 • Covalent Bonding Single Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it

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is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared electrons.

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Chapter 8: Covalent Bonding  
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Covalent bonds are formed when electrons are shared between atoms. Only non-metal elements participate

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in covalent bonding. As in ionic bonding, each atom gains a full outer shell of electrons. Covalent bonds between atoms are strong and the bonded unit of atoms is what is called a molecule.



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1) Ionic bonds-transfer of electrons. (gained or lost; makes formula unit)

2) Covalent bonds-sharing of electrons. The resulting particle is called a

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“molecule”. Covalent Bonds.

- The word covalent is a combination of the prefix co- (from Latin com, meaning “with” or “together”), and the verb. valere, meaning “to be strong”.

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Chapter 8 Covalent Bonding  
Answers Key

In the Covalent Bonding  
game, players bond together  
atoms to create target  
molecules with unique bond

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polarities and molecular shapes. Check out our Free Teacher Resources here! Take a peek inside the Covalent Bonding game for a brief overview of the concepts covered through gameplay.

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Collisions: Covalent Bonding  
game - Playmada Games

The animated PowerPoint  
takes students through the  
basics of covalent bonding.  
It contains examples, tasks  
and answers. I've found this

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to be a very useful resource. It clicks through each stage separately and so you can click at the pace your students need.

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electronegativities, classify the bonding in each compound as ionic, covalent, or polar covalent. Indicate the direction of the bond dipole for each polar covalent bond. NaBr; OF<sub>2</sub>; BCl<sub>3</sub>; the S-S bond in CH<sub>3</sub>

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CH<sub>2</sub>SSCH<sub>2</sub>CH<sub>3</sub>; the C-Cl  
bond in CH<sub>2</sub>Cl<sub>2</sub>; the O-H  
bond in CH<sub>3</sub>OH; PtCl<sub>4</sub><sup>2-</sup>

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Chapter 8 Covalent Bonding

183 Section Review

Objectives • State a rule  
that usually tells how many  
electrons are shared to form  
a covalent bond • Describe

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how electron dot formulas are used • Predict when two atoms are likely to be joined by a double or a triple covalent bond • Distinguish between a single covalent bond and other covalent bonds

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