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0.500 M NaOH To Titrate It?
2) A Student Weighs Out
1.051 g Of Impure KHP,
Dissolves The Sample In
Deionized Water And Titrates
It With 0.1101 M NaOH

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Determination of the
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Weight of KHP 0.1149 0. 7049
0.119 Starting Buret
Reading: 1. 74 ml/ 1.01ml
3.05ml Ending Buret Reading:
37-08 ml 36.37 m 38.45 ml mL
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Department of Chemistry

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Approach through an

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following laboratories have
been compiled and adapted by
Alan Khuu, M.S. & Armando
Rivera, Ph.D.

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Manual, 2017 Revision

Questions: 1) What mass of KHP would you need in order to use 35.00 mL of 0.400M NaOH to titrate it? 2) A student weighs out 1.000 g of impure KHP, dissolves the sample in deionized water and titrates it with 0.1101 M NaOH solution.

~~Answered: 105 Chemistry I
Laboratory Manual, 2017... |
bartleby~~

This General Chemistry Laboratory manual consists of experiments designed to introduce to a student new to chemistry basic chemical concepts and techniques. Each lab consists of an introduction, where the

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major concepts of the experiments are explained, safety instruction, and a step-by-step procedure. The major learning outcomes emphasized in this lab manual are as followed:

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Practice Exam III Key B.
Practice Final Exam
Problems. PF answers pg1-6.
CH141 Practice Final Key II
(pages 6-12) CH141 Exam I
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