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Atomic Structure Worksheet Answers Chemistry

Atomic Structure Worksheet – Solution Key. Fill in the blanks for the elements in this chart. For the purposes of this chart, round all atomic masses to the nearest whole number. Element Number of Protons Number of Neutrons Number of Electrons Atomic Mass Atomic Number. lithium 3 4 3 7 3.

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Chemistry a study of matter 3.6 answer key. Isotopes are atoms of the  
same element with different numbers of neutrons which results in  
different mass numbers.

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## Chemistry A Study Of Matter 3.6 Answer Key

On our dedicated atomic structure page you will find all the resources you require to revise the history of the atom and the development of the atomic model including electronic structure. Whether you are looking for AQA GCSE Chemistry revision materials for atomic structure or you are studying combined science, the worksheet and resources on ...

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www.njctl.org Chemistry Atomic Structure Answers 1. In the nuclear  
model of the atom, protons (and neutrons) are housed in a small,  
dense nucleus. Electrons surround the nucleus in an area of mostly  
empty space. 2. If electrons are electrically attracted to nucleus and  
would, therefore, have centripetal acceleration in order to orbit the  
nucleus.

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## Atomic Structure: Chapter Problems

Question 6. Define the following terms: (a) Atomic number (b) Mass  
number (c) Nucleons (d) Valence shell Answer: (a) Atomic number:  
Atomic number refers to the number of protons present in an atom. It  
is denoted by Z. Example: An atom of oxygen contains 8 proton

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Therefore its atomic number is 8.

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Name Datu Directions: read and answer each question in order to receive credit (1) Indicate whether each of the following is a permissible set of quantum numbers. If the set is not permissible, state why it is not a.  $2, 4, 0, m = +1$  b.  $n = 5, 3, m = -3$  c.  $n = 4, l = 3, m = 2$  d.  $n = 0, l = 0, m = 4$  (2) Indicate whether each of the following is a permissible set of quantum numbers.

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