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Chemistry Live! | Worked Solutions 76 Workbook Chapter 9 | The Mole Concept W9.1 (a) 1 mole of Li atoms = 7 g (b) 1 mole of Na atoms = 23 g (c) 1 mole of Ca atoms = 40 g (d) 1 mole of Fe atoms = 56 g (e) 1 mole of Ag atoms = 108 g (f) 1 mole of Pb atoms = 207 g W9.2 (a) 1 mole of Cl atoms = 35.5 g 10 moles of Cl atoms = 35.5 × 10

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Chemistry Live! | Worked Solutions 2 Chapter 9 - The Mole Concept 9.1 Since a sulfur atom (Ar = 32) weighs twice as much as an oxygen atom (Ar = 16), 32 g of sulfur atoms and 16 g of oxygen atoms both contain the same number of atoms.

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