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Classification and nomenclature |

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10 Haloalkanes And Haloarenes

Haloarenes are less polar than haloalkanes and are insoluble in water. This is because of lack of hydrogen bonding. As a result, the attractive forces in haloarenes—water system remain less than the attractive forces in H₂O molecules which are hydrogen bonded.

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Chemistry Chapter 10 ...

Chemistry Chapter-10: Haloalkanes and Haloarenes Question 10.1: Name the following halides according to IUPAC system and classify them as alkyl, allyl, benzyl (primary, secondary, tertiary), vinyl or aryl halides :

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Notes Chapter 10 The bond (carbon-halogen bond) of alkyl halide is polarised. Halogen atom carries a partial negative charge Carbon atom carries a partial positive

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Notes Chemistry Chapter 10. 1.
Haloalkanes are classified as fluoro, chloro, bromo or iodo compounds according to the type of halogen present and as mono-, di- tri-, tetra-haloalkanes, etc., according to the one, two, three, four, etc., halogen atoms respectively present in their molecule.
2.

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Haloalkanes are the halogen derivatives of hydrocarbons and these are of the following types depending upon the number of hydrogen atoms present in them. Haloarenes are aromatic compounds in which the halogen atom is attached directly to the carbon atom of the aromatic ring. The solutions of chapter 10 also give exemplary problems, HOTS (Higher

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alkyl halides: X is more
electronegative than carbon. So, the C-
X bond is polarized with C having a

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haloarenes
partial positive charge and X having a partial negative charge. Preparation of haloalkanes:

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Preparation of Haloalkanes I: 2:

Preparation of Haloalkanes II: 3:

Properties of Haloalkanes I: 4:

Properties of Haloalkanes II: 5:

Properties of Haloalkanes III: 6: Alkyl

Halides and Dihalides I: 7: Reactions

Of Dihalides I: 8: Preparation of

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