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Chemical Reactions Vocabulary

Product. A new substance formed in a chemical reaction. **Reactant.** A starting substance in a chemical reaction.

Chemical Equation. A concise representation of a chemical reaction.

Balanced Equation. An equation in which each side has the same number

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a solid that forms and settles out of a liquid mixture; form in double replacement. combustion reaction. a

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Chemical change in which an element or a compound reacts with oxygen, often producing energy in the form of heat and light; requires oxygen; often with hydrogen and carbon, often produces carbon dioxide and water.

Chapter 11 - Chemical Reactions

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Chemistry Chapter 11 Vocabulary of the entire chapter. Terms in this set (14) Chemical Equation. An expression representing a chemical reaction; the formulas of the reactants (on the left) are connected by an arrow with the formulas for the products (on

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the right) Skeleton Equation. A chemical equation that does not indicate the relative amounts of reactants and products.

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Chem chapter 11 vocab Stoichiometry:

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The study of quantitative relationships between the amounts of reactants used and products formed by a chemical reaction; is based on the law of conservation of mass. mole ratio: In a balanced equation, the ratio between the numbers of moles of any two substances. limiting reactant: A

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reactant that is totally consumed during a chemical reaction, limits the extent of the reaction, and determines the amount of product.

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Crossword Down: 1) A chemical change that involves an exchange of positive ions between two compounds.
2) A chemical equation that does not indicate the relative amounts of the reactants and products.

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Vocabulary Product. A new substance
formed in a chemical reaction.

Reactant. A starting substance in a
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A concise representation of a chemical reaction. Balanced Equation. An equation in

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Chapter 11 "Describing Chemical Reactions". Use these activities to

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study the vocabulary and major concepts presented in this chapter. a representation of a chemical reaction with formulas of reactants to the left of an arrow and the formulas of the products to the right of an arrow.

Quia - Chapter 11 "Describing

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a representation of a chemical reaction, the formulas of the r... a chemical substance that is present at the start of a chemica... A substance produced in a chemical reaction. a chemical equation that does not indicate the relative amount...

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a chemical reaction that occurs when a substance reacts with oxygen, releasing energy in the form of heat and light synthesis reaction A chemical reaction in which two or more simple

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reactants combine to produce a single product.

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Key Terms Balance equations How to write a skeleton equation How to write a word equation

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A chemical change in which two or more substances react to form a single new substance. Combustion reaction.

A chemical change in which an element or a compound reacts with

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oxygen, often producing energy in the form of heat and light. Complete ionic equation.

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Chapter 11 "Describing Chemical Reactions" Use this activity to test

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Your knowledge of the vocabulary and major concepts presented in this chapter

Quia - Chapter 11 "Describing Chemical Reactions"

a compound reacting with an acid to form a salt and water. boiling point. the

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temperature at which a liquid boils at sea level. catalyst. substance that initiates or accelerates a chemical reaction. chemical. material produced by changes in atoms or molecules. chemical bond. an electrical force linking atoms.

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330 Chapter 11 11.2 Types of
Chemical Reactions Often charcoal
briquettes provide the heat for
barbeque grills through the burning of

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carbon. Have you ever felt the heat and smelled the smoke coming from a burning charcoal grill? The heat and smoke are the products of a combustion reaction. Combustion is one of the 4ve general types of chemical

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11.2 Types of Chemical Reactions - bleiker.weebly.com

Chemical reaction - Process by which the properties and composition of matter change

1. Chemical change - properties and composition and of matter change
2. Physical change - properties change, but not the

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Vocabulary Review 4. Reactant - a substance that is present at the start of a chemical reaction 5.

**producing Carbon dioxide (CO -
Campbell County Schools**

Salt: obtained by evaporating water.

Oxidation-reduction reaction: a

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reaction that involves a transfer of electrons. Precipitation Reaction: a reaction where the formation of a solid when two solutions are mixed occurs. Double-Displacement Reaction: a reaction in which the reactants are double exchanged.

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... A representation of a chemical reaction that uses symbols to show the relationship between the reactants and the products. 10 Reactant A substance

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or molecule that participates in a
chemical reaction. 11 Product The
substance that forms in a chemical
reaction. 12 Law of ...

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