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1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify the solvent in each of the following examples:

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Modern Chemistry Chapter 12 Mixed CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl_2 does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl_2 is an ionic solid. In the crystal form, its ions are locked in position. Dissolving CaCl_2 in water

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