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ch 12, lec 1, Exercise 12  
Theorem no 1 - 10th Class  
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### of Reaction \u0026amp; Equations

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*Answer - Matric Part 2*

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Solutions Section~~

Modern Chemistry 5 Solutions

CHAPTER 12 REVIEW Solutions

SECTION 2 SHORT ANSWER

Answer the following

questions in the space

provided. 1. The following

are statements about the

dissolving process. Explain

each one at the molecular

level. a. Increasing the

pressure of a solute gas

above a liquid solution

increases the solubility of

the gas in the liquid.

~~CHAPTER 12 REVIEW Solutions~~

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A homogeneous mixture that contains particles in a dispersed phase that do not settle out is a solution. The solution process with water as the solvent. hydration. Liquid solutes and solvents that are not soluble in each other are immiscible. A solute molecule that is surrounded by solvent molecules is solvated.

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Chapter 12 Solutions review.  
STUDY. PLAY. solution.  
homogeneous mixture of two or more substances,  
consisting of ions or

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molecules. solute. gas or solid dissolved in a liquid or solution component in small amount. solvent. liquid that dissolves a gas or solid or the solution component in greater amount.

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CHAPTER 12 REVIEW Solutions  
SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2.



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## Review Solutions Section 3

Identify the solvent in each of the following examples:

~~me06se eFMsr i vi~~

Chapter 12 Review Solutions

- 1) 56 2) = 4, = 10, = 16, = 22, = 28 3) = 4, = 8, = , = 64, = 4)  $an = 2n - 3$  5) = •  
4n 6) = 5, = 3, = 1, = -1 7)  
 $3 + 4 + 5 + \dots + (n + 2)$  8)  
9) 15 10)  $d = -5$ ; = 4, = -1,  
= -6, = -11 11) =  $94 - 10n$ ;  
= 14 12) 220 13) = 73,  $d =$   
 $-5$ , = - 5 14) 34980 15) 25

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~~, = 10, = 16, = 22, = 28 +  
1) = 2.~~

CHAPTER . 12 . REVIEW ' ' ' --  
Solutions . SHORT ANSWER

Answer the following  
questions in the space

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## Review Solutions Section 3

provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

~~CHAPTER REVIEW Solutions~~

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## CHAPTER 12 REVIEW Solutions SECTION 3 SHORT ANSWER

Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. \_\_\_\_\_

\_\_\_\_\_ b. Mary needs a 0.02 M solution of  $\text{NaNO}_3$ . She calculates that she needs 2.00

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This Chapter 12 Review,  
Section 1: Solutions  
Worksheet is suitable for  
9th - 12th Grade. Here is a  
different approach to  
solutions: a worksheet that  
has chemistry learners use  
words to describe them! This  
is an outstanding exercise  
that will stretch learners  
to show what they know. They  
answer questions comparing  
solutions, suspensions, and  
colloids.

~~Chapter 12 Review, Section  
1: Solutions Worksheet for  
9th ...~~

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## Chapter 12 Review Problems

INSTRUCTIONS: Write your answers on lined or blank sheets of paper. You do not need to write the question, ONLY WRITE THE PROBLEM NUMBER and ANSWERS/SOLUTIONS. • For problems that involve calculations, you must show your work to get full credit.

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Modern Chemistry Chapter 12  
Solutions Review Answers  
Modern Chemistry Chapter  
12-Solutions Section 1-  
Types of Mixtures Solutions  
are homogeneous mixtures of  
two or more substances in a  
single phase. Soluble

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describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution.

~~Chapter 12 Review Solutions~~

~~Section 3 Answers — Kora~~

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~~12-Solutions Section 1-~~

~~Types of Mixtures Solutions~~

~~are homogeneous mixtures of~~

~~two or more substances in a~~

~~single phase. Soluble~~

~~describes a substance as~~

~~capable of being dissolved.~~

~~Solvent is the dissolving~~

~~medium in a solution. Solute~~

~~is the substance that is~~

~~dissolved in a solution.~~

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Matter. Explain your answer.

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