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### Ch. 28: Nuclear Chemistry

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These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

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## RADIOCHEMISTRY AND NUCLEAR CHEMISTRY □ Vol. I -

Radiochemistry and Nuclear Chemistry - Sándor Nagy

©Encyclopedia of Life Support Systems (EOLSS) The role of chemistry in nuclear science (and related fields) can be seen in Figure 1 showing that there were 10 such years in the 105-year history of Nobel Prizes when the

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Chemistry. The composition of the nucleus changes only during nuclear reactions. The 3 ways to change nuclear composition in an attempt to become stable? Radioactive decay, Fission, and Fusion.

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### Chapter 25 Nuclear Chemistry Guided Reading Answers

Chemistry End of Chapter Exercises. Write a brief description or

definition of each of the following: (a) nucleon (b)  $\alpha$  particle (c)  $\beta^-$

particle (d) positron (e)  $\gamma$  ray (f) nuclide (g) mass number (h) atomic

number. Which of the various particles ( $\alpha$  particles,  $\beta^-$  particles, and so on) that may be produced in a nuclear reaction are actually ...

### 21.2 Nuclear Equations □ Chemistry

Nuclear fusion: It is a type of reaction where two or more elements fuse together to form a larger element, releasing a large amount of energy in the process. To learn more about nuclear chemistry topics

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CHAPTER 3 - Nuclear Mass and Stability. Pages 41-57. Select CHAPTER 4 - Unstable Nuclei and Radioactive Decay. ... chapters from the broad introduction through the more in depth descriptions of radiochemistry then nuclear radiation chemistry and finally the guide to nuclear energy (including energy production, fuel cycle, and waste management).

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