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CHAPTER 8 REVIEW Chemical Equations and Reactions SECTION 2

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SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. c synthesis (a) $AX + BY \rightarrow AY + BX$ d decomposition (b) $A + BX \rightarrow AX + B$ b single-displacement (c) $A + B \rightarrow AX$ a double-displacement (d) $AX \rightarrow A + X$

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8 Chemical Equations and Reactions

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mainly known as reactivity.

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Answers
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Chemical Reactions

Answers Answer Key. Chemical reaction. The process by which one or more substances are changed into one or more different substances. Reactant. represents, with symbols and formulas, the identities of relative molecular or molar amounts of the reactants and products in a chemical reaction.

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Chapter 8 Chemical Reactions Answer
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8: Quantities in Chemical Reactions

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evolution of energy as heat and light(has
to be both), many physical changes also
involve either heat or Guidelines for

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balancing chemical equations. Balance the different types of atoms one at a time.

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Possible answer: The decomposition reaction of water requires energy to break down the water molecules into hydrogen and oxygen. If enough energy is not added, the reaction will not occur. The instrument the teacher used was not

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Answers
passing enough energy into the water for it to decompose.;

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chapter 8: chemical reactions 8.3

balancing chemical equations balancing

suggestions : 1) count the #. Types of

Reactions Worksheet Indicate the type of

reaction taking place: 1) $\text{NaBr} + \text{H}_3\text{PO}_4$

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Answers
Na₃PO₄ + HBr Reaction type: 2)

Ca(OH)₂ + Al₂(SO₄)₃ ..

Types Of Reactions Homework Answer
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Types of chemical reactions
Chapter

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Chemical Reactions

8.2 Objectives
Define and give general equation for synthesis, decomposition, single-replacement, and double-replacement reactions.
Classify a reaction as synthesis, decomposition, single-replacement, double-replacement, or combustion.
List three types of synthesis reactions and six

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types of decomposition reactions.
List four types of single-replacement reactions and three types of double-replacement reactions.<br ...

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Answers
months ago. by mrscarbol. Played 76 times. 0. 10th grade . Chemistry. ... none of the answers are correct. Tags: Question 16 . SURVEY Which chemical reaction takes place when 2 substances react to form a single product? answer choices

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Matter cannot be created or destroyed in an ordinary chemical reaction. Mass of reactants must equal mass of products. Equations must be balanced because no atoms can be gained or lost. a) solid . b) liquid . c) gas . d) aqueous solution

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(substance is dissolved in water)

Answer Key: Chemical Reactions Review
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Chemical Reactions

Answers
Chemical reactions are described by chemical equations. A chemical equation represents, with symbols and formulas, the identities and relative molecular or molar amounts of the reactants and products in a chemical reaction. For example, the following chemical equation shows ...
reaction CHAPTER 8 ...

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CHAPTER 8 Chemical Equations and
Reactions

Chapter 8 - Reaction Equations & Types

All answer keys are for the WS in the
Packet: ... Chapter 8 Reactions. Chapter 9
Stoichiometry. Chapter 10-11 Gases.
Chapter 12-13 Solutions & Properties.

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