

## Chemistry Balancing Equations Worksheet Answer Key

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49 [Balancing Chemical Equations Worksheets \[with Answers\]](#)  
[Balancing Chemical Equations with Odd Number of Atoms on Elements](#). Another area wherein balancing becomes a tricky affair is during the presence of odd subscripts or atoms of an element. Let us take into consideration, this particular equation:  $\text{NH}_3 + \text{O}_2 \rightarrow \text{NO} + \text{H}_2\text{O}$

100 [Balancing Chemical Equations Worksheets with Answers ...](#)  
[Balancing an unbalanced equation](#) is mostly a matter of making certain mass and charge are balanced on the reactants and products side of the reaction arrow. This is a collection of printable worksheets to practice balancing equations. The printable worksheets are provided in pdf format with separate answer keys.

[How to Balance Equations - Printable Worksheets](#)  
[Balancing Chemical Equations Answer Sheet](#). Displaying top 8 worksheets found for - Balancing Chemical Equations Answer Sheet. Some of the worksheets for this concept are Balancing equations practice problems, Teacher answer balancing equations, Balancing chemical equations answer, Chemical formulas equations work answers, Another balancing equation answer key, Balancing equations work answers, Balancing word equations chapter 9, Chapter 7 work 1 balancing chemical equations.

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[Balancing Word Equations Worksheet Answer Key Worksheet ...](#)  
[Balancing Equations Worksheet and Key 1](#). Answer the following questions about the chemical equation shown below:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  a) What are the reactants? H 2 and O 2 b) What is the product? H 2 O c) What is the number "2" in front of the H 2 (and H 2 O) called? Coefficient d) Is the reaction balanced?

[Balancing equation - Balancing Equations Worksheet and ...](#)  
[Balancing Chemical Equations - Answer Key](#) Balance the equations below: 1)  $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  2)  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  3)  $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$  4)  $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$  5)  $1\text{Pb(OH)}_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2$  6)  $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2(\text{SO}_4)_3$  7)  $1\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O}$  8)  $1\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$  9)  $2\text{C}_8\text{H}_{18} + 25\text{O}_2 \rightarrow 16\text{CO}_2 + 18\text{H}_2\text{O}$  10)  $1\text{FeCl}_3 + 3\text{NaOH} \rightarrow 1\text{Fe(OH)}_3 + 3\text{NaCl}$  11)  $4\text{P} + 5\text{O}_2 \rightarrow 2\text{P}_2\text{O}_5$

[Balancing Chemical Equations - Mr. Kent's Chemistry ...](#)  
[Balancing Chemical Equations Worksheet 1 - ANSWERS](#) 1.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  2.  $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$  3.  $\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$  4.  $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$  5.  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  6.  $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$  7.  $2\text{Li} + \text{F}_2 \rightarrow 2\text{LiF}$  8.  $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$  9.  $2\text{NH}_3 + 3\text{N}_2 + 2\text{H}_2 \rightarrow 10\text{H}_2\text{N}_2$  10.  $2\text{HCl} + \text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$  11.  $2\text{NI}_3 + \text{N}_2 + 3\text{I}_2 \rightarrow 2\text{HI}$  12.  $2\text{HI} + \text{H}_2 + \text{I}_2 \rightarrow 4\text{HI} + 3\text{O}_2$  13.  $2\text{AI} + 2\text{O}_3 \rightarrow 3\text{I}_2$

[Balancing Chemical Equations Worksheet 1](#)  
Worksheet with various chemical equations for pupils to balance. Answers included too.

[Balancing Chemical Equations | Teaching Resources](#)  
A worksheet aimed at GCSE and A level chemistry students which focuses on balancing equations. The following premium resource for balancing equations is available in our shop which also includes worked answers: We have hundreds of chemistry resources in our green APL Shop <https://www.tes.com/teaching-resources/shop/greenAPL> covering topics such as atomic structure, kinetics, equilibria, bonding, mole calculations, electronic configuration, group 1 elements, group 7 elements, redox ...

[FREE Chemistry: Balancing Equations | Teaching Resources](#)  
[Balancing Equations: Answers to Practice Problems](#). 1. Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers). (a)  $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$ . (b)  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$ . (c)  $2\text{FeBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 6\text{HBr}$  (d)  $1\text{C}_4\text{H}_{10} + 13\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{C}_2\text{H}_6\text{O}$ .

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If you can balance these, you can balance any equation given in class. 1.  $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$  2.  $\text{Al}_2(\text{SO}_4)_3 + \text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + \text{Al(OH)}_3$  3.  $\text{Al}_2\text{O}_3 + \text{Fe} \rightarrow \text{Fe}_3\text{O}_4 + \text{Al}$  4.  $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$  5.  $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \text{H}_2\text{O}$  6.  $\text{NaHCO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O} + \text{CO}_2$  7.

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[Balancing Chemical Equations Practice Worksheet with ...](#)  
[Word Equations Worksheet](#) Write and balance the following chemical equations. 1) When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made. 2) When isopropanol (C3H8O) burns in oxygen, carbon dioxide, water, and heat are produced.

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