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#25 *Intermolecular Forces and Boiling*

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anabug00.Chemistry Chapter 11. organic
compounds. number of carbon atoms for

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~~Answers~~ number of carbon atoms for methane. number of carbon atoms for ethane. number of carbon atoms for propane.-The formulas of organic compounds are written with carbon fir....

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Chapter 11 187 Exercise 11.3 – Equation Stoichiometry: Iron is combined with carbon in a series of reactions to form pig iron, which is about 4.3% carbon. $2C + O_2 \rightarrow 2CO$ $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$ $2CO + C \rightarrow 2CO_2$ (in iron) CO_2 Pig iron is easier to shape than pure iron, and the presence of carbon lowers its melting point

~~Chapter 11 – Gases~~

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~~Answers~~ Electron Configurations To get a review of the most important topics in the chapter, fill in the blanks in the Key Ideas section. Work all of the selected problems at the end of the chapter, and check your answers with the solutions provided in this chapter of the study guide. Ask for help if you need it.

~~Chapter 11 Modern Atomic Theory—An Introduction to Chemistry~~

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Chapter 1 Basic Concepts of Chemistry
and Chemical Calculations; Chapter 2
Quantum Mechanical Model of Atom;
Chapter 3 Periodic Classification of
Elements; Chapter 4 Hydrogen; Chapter 5
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6 Gaseous State; Chapter 7
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FAQ. Q: What are hydrophobic and hydrophilic? A: Hydrophobic is a “fear of water”. Nonpolar substances are typically hydrophobic. They are not attracted to water. Hydrophilic on the other hand are “water lovers”. These compounds are attracted to water, and therefore are commonly ionic or polar.

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