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Chemical Equations Worksheet
Write the balanced chemical
equation (excluding
spectator ions) underneath
each reaction description,
and answer the question.

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Equations Worksheet

Answer: Hydrogen gas is released at the cathode (the negative electrode). (o) A solution of copper(II) sulfate is electrolyzed using inert electrodes.

$2\text{Cu}^{2+} + 2\text{H}_2\text{O} \rightarrow 2\text{Cu} + \text{O}_2 + 4\text{H}^+$ Is the final solution acidic, basic, or neutral?

Answer: Acidic hydrogen ions (H^+) are produced in the reaction. (p) A solution of ammonium sulfate is added to a satu-

CF#10826 AP Chemistry

Chemical Equations Worksheet-

Rev

1 $\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O}$
2 $2\text{Na}^+ + 2\text{Cl}^- \rightarrow 1\text{NaCl}$
3 $4\text{Al} + 3\text{O}_2$

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Name: Date: Balancing
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Whether a solution

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A chemical equation
describes what happens in a
chemical reaction.The

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Answer Key identifies the reactants (starting materials) and products (resulting substances), the formulas of the participants, the phases of the participants (solid, liquid, gas), the direction of the chemical reaction, and the amount of each substance.

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Sweet 16 Chemistry of Gases
Tournament continued 2 21
linn Scientific nc ll ihts
esere Sweet 16 Chemistry of
Gases Tournament
Supplementary Chemical
Equations Worksheet Predict

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the products and balance the following chemical equations to identify the contestants in the Chemistry of Gases tournament. 1. Na 2 CO 3 (s) + ___ HCl(aq)?

Sweet 16 Chemistry of Gases Tournament SCIENTIFIC - Flinn

In a balanced chemical equation, both the numbers of each type of atom and the total charge are the same on both sides. Equations 4.1.1 and 4.1.2 are balanced chemical equations. What is different on each side of the equation is how the atoms are arranged to make molecules or ions.

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4.1: Chemical Reactions and Chemical Equations - Chemistry ...

Chemistry Chemical Equations
Flinn Answer Key To balance
a chemical equation, enter
an equation of a chemical
reaction and press the
Balance button. The balanced
equation will appear above.

Chemistry Chemical Equations Flinn Answer Key

Step 1, Write down your
given equation. For this
example, you will use: C_3H_8
 $+ O_2 \rightarrow H_2O + CO_2$ This
reaction occurs when propane
(C_3H_8) is burned in the
presence of oxygen to
produce water and carbon
dioxide. Step 2, Write down

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the number of atoms per element. Do this for each side of the equation. Look at the subscripts next to each atom to find the number of atoms in the equation. When writing it out, it's a good idea to connect it back to the original equation, noting how each element ...

How to Balance Chemical Equations: 11 Steps (with Pictures)

Replace immutable groups in compounds to avoid ambiguity. For example, $C_6H_5C_2H_5 + O_2 = C_6H_5OH + CO_2 + H_2O$ will not be balanced, but $XC_2H_5 + O_2 = XOH + CO_2 + H_2O$ will. Compound states [like (s) (aq) or (g)] are

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Answers Key. You can use parenthesis () or brackets [].

Chemical Equation Balancer

Every balanced chemical equation consists of two parts: the reactant side and the product side. Both of these sides are separated by the means of an arrow. On the left side of the arrow, you will find the reactant side. This side represents the elements which are used for initiating the experiment.

100 Balancing Chemical Equations Worksheets with Answers ...

To get four hydrogen atoms

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Answer Key, add a coefficient of 2 for the hydrogen gas. The coefficient is a number that goes in front of a chemical formula. Remember, coefficients are multipliers, so if we write $2 \text{ H}_2 \text{ O}$ it denotes $2 \times 2 = 4$ hydrogen atoms and $2 \times 1 = 2$ oxygen atoms. $\text{SnO}_2 + 2 \text{ H}_2 \text{ O} \rightarrow \text{Sn} + 2 \text{ H}_2 \text{ O}$.

3 Steps for Balancing Chemical Equations -

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Equations Worksheet AP
Chemistry Chemical Equations
Worksheet Write the balanced
chemical equation (excluding
spectator ions) underneath

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Answer Key
each reaction description,
and answer the question. (a)
A solution of ammonia is
added to a dilute solution
of acetic acid. Identify the
conjugate acid-base pairs in
this reaction.

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Images Great job! You did so
well on this quiz you could
tutor others on how to
balance equations! If you're
feeling a bit shaky on all
the steps and details, you
can review the simple method
of balancing
equations. Otherwise, you may
wish to review how to

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Answer Key
balance oxidation-reduction
or redox reactions or move
on to understanding mole
relations in balanced
equations.

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