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1 ( Chem 1 H) GCSE

Science Revision

Chemistry \ "The Three

States of Matter \ " The

Creation of Chemistry -

The Fundamental Laws:

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#3 Matter and Change

~~Chapter 1~~

~~Introduction: Matter and~~

~~Measurement~~ States of

matter | States of matter

and intermolecular

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Matter and the Particle

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~~and Change Part 1~~

Particulate nature of

matter - Part-1 -

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States of Matter

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Chemistry Physical  
Changes

---

States of Matter : Solid  
Liquid Gas10 Amazing  
Experiments with Water

---

Answers

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Pure Substances and  
Mixtures!

(Classification of  
Matter)Physical and  
Chemical Changes

---

Understanding Atoms,  
elements, and molecules  
Part #1 (9min)

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1.2 What is matter? -

HD Properties of Matter

Physical and Chemical

Changes: Chemistry for

Kids - FreeSchool Phase

Changes States of

Matter Solids, Liquids,

Gases \u0026 Plasma

Chemistry Change of

State | Matter | Physics |

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Matter and Changes of

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States of Matter \u0026amp;

Changing State #20

Pure Substances and

Mixtures, Elements

\u0026amp; Compounds.

Classification of Matter.

Chemistry Examples.

Chemistry/ICSE/Class

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~~Classifying Matter With~~

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Chemicals Ozone Layer,  
atmosphere, ozone  
formation,

chlorofluorocarbons,  
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and Matter Chemistry  
Central Science

Chapter Summaries □

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Chapter 1 Section 2

MATTER and Its

Properties Matter

Anything that has

MASS and VOLUME.

ATOMS and

MOLECULES the basic

building blocks of

matter. ATOM-- the

smallest unit of an

element that still

maintains the properties

that element.

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**ELEMENT** a pure substance made up of only one type of atom.

**MATTER** and  
**CHANGE** Chapter 1  
Section 1

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Change Chapter 6 Study  
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**CHAPTER** Class

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atomic radius and ionic

radius. Circle the letter

of the choice that best

completes the statement

or answers the question.

1. Atomic radii cannot  
be measured ...

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CHAPTER 3

SOLUTIONS

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Matter Properties and C

hanges Matter Properties

and Changes Solutions

Manual Chemistry:

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Chapter 3 35 Section 3.1

Properties of Matter

pages 70□75 Problem-

Solving Lab 1. Explain

why the flow of a

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compressed gas must be controlled for practical and safe use.

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Matter Properties and Changes  
Matter Properties and Changes

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Periodic Law Section  
6.1 Development of the  
Modern Periodic Table  
1. octaves 2. eight 3.  
nine 4. accepted 5.  
Dmitri Mendeleev 6.  
atomic mass 7. elements  
8. atomic number 9.  
Henry Moseley 10.  
protons 11. periodic law  
12. properties 13. 14 ...

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answers

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Chapter 5 25 Electrons

in Atoms Reviewing

Vocabulary Match the

definition in Column A

with the term in Column

B. Column A Column B

1. The set of frequencies

of the electromagnetic

waves emitted by the

atoms of an element 2.

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The minimum amount of energy that can be lost or gained by ...

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study guide Chemistry

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numbers used in

chemistry can be  $q$  the  
study of energy changes  
(that is, energy

produced or absor $\square$  A  
system that can

exchange neither energy

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extensive property.

intensive property.

extensive properties

(examples) intensive

property (examples) a

property that depends on

the amount of matter in

a sample. a property that

depends on the type of

matter in a sample not□.

mass, volume, length,

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weight.

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Transparency Masters

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number of moles of  
known substance C H C  
H C H O O O CO CO  
CO H O H O H O

number of moles of

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unknown substance O

CO H O C H CO H O C

H O H O C H O CO use

mole ratio 5 mol O 2 1

mol C 3 H 8 3 mol CO 2

1 mol C 3 H 8 4 mol H

2 O 1 mol C 3 H 8 1

mol C 3 H 8 5 mol O 2

MATH SKILLS

TRANSPARENCY

MASTER 16

A [chemical change] is  
a change that produces

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matter with a different composition than the original matter.

Chemical Change A

change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of

Compounds Quite

different properties than their component

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