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Chemistry 1003: Molarity and Colligative Properties ...

Note Taking Guide: Episode 605 Name: Camila Paredes Date: 3-20-2020 Period: 1 CHEMISTRY: A Study of Matter © 2004, GPB 6.27 Descriptive Abbreviations (S) Solid (l) Liquid (g) Gas (aq) aqueous Determining State at Room Temperature • All metals are solids except for mercury, which is a liquid.

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Note Taking Guide: Episode 501. chemical bond. ionic bond. cation. anion. forms when 2 or more atoms rearrange valence electrons to incr.... forms when valence electrons are transferred from one atom to a.... atom loses electrons to become positively charged. atom gains electrons to become negatively charged.

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Note Taking Guide: Episode 304 Name Aditi Acharya Quantum Numbers N, l, m, s Used to describe an electron in an atom n Principle quantum number Represents main energy level of electron maximum # of electrons in an Energy level = $2n^2$ Example: What is the maximum number of electrons that can be in the 5th main energy level? l The 2nd quantum number • Describes the orbital

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shape within an energy level • number of orbital shapes possible in energy level = n Orbital Shapes designated s,p ...

note_ep304aditi.pdf - orientation of orbital in space s ...

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give away electrons, ions are positively charged, cations. nonmetals. take electrons, ions are negatively charged, anions. writing ionic formulas. 1. write the symbols for the element. -the cation is written first. -The anion is written second. 2. Determine the charge on each ion.

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Note Taking Guide: Episode 102 Name: Nayan Jacquet

International System (SI) Measuring system used in science AKA metric system Length - distance between two points Base unit- meter (M) Instrument- meter stick or metric ruler Mass -amount of

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matter in an object. Base unit- gram (G) Instrument- balance
Volume -amount of space an object occupies Base unit-Liter (L)
Instrument- metric ruler or graduated cylinder Time - interval
between two occurrences Base unit- seconds (S) Instrument ...

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Study of Matter © 2004, GPB 7.1 Lab results: 1 doz grains of rice =
_____g (Use this fact as a conversion

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