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12. Inert gasses are arranged in what part of the Periodic Table? Far right group or column. PART III - Chemical Reactions, Reactants and Products. 13. A liquid that forms a gas when heated is an example of a chemical reaction. False 14. When Vinegar (acetic acid  $\text{CH}_3\text{COOH}$ ) reacts with baking soda ( $\text{NaHCO}_3$ ), one of the products is Carbon Dioxide ( $\text{CO}_2$ ). True 15.

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objective was to determine the molar mass of the volatile liquid. Volatile liquids are molecular substances that have very low boiling points and small masses. . Molar masses are among the first ...

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molecule on the reactant side. Because we added HCl and NaOH in steps 1-4, the colours changed back to their original shades to maintain equilibrium. 7)  $\text{BaCr}_2\text{O}_7$  is much more soluble than  $\text{BaCrO}_4$ , as the  $\text{K}_2\text{Cr}_2\text{O}_7$  solution was darker in colour when mixed with  $\text{Ba}(\text{NO}_3)_2$ . 8) No precipitate formed in step 7 of Part IV because H in HCl was added,



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