

## Chemquest 19 Answer Key

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19. ChemQuest 6. Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Specific Heat Constants Depend on the State of Matter . Recall that specific heat is the energy necessary to raise the temperature of one gram of a substance. by one degree Celsius. The specific heat of liquid water is a constant equal to 4.18 J/(g·°C). This means that it takes 4.18 Joules (J) of heat energy to raise the ...

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ChemQuest 10 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Bohr 's Solar ... 11. The energy difference between the 1st and 2nd energy levels in a certain atom is 5.01 x 10<sup>-19</sup> J. What frequency of light is necessary to excite an electron in the 1st energy level? 12. a) What is the frequency of light given off by an electron that loses 4.05 x 10<sup>-19</sup> J of energy as it moves from the 2nd to the ...

~~Critical Thinking Questions: Bohr 's reasoning~~

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Created Date: 10/19/2012 9:41:11 AM ChemQuest 8: Answer Key An ion with a negative charge. How is an anion formed? An anion is formed by adding electrons to an atom. How is an anion named? An anion is named by replacing the suffix of the element name with “ -ide ” followed by the word ion. What elements on the period table tend not to form ions? Why? Noble gases. CHEMQUEST 18 FORMING IONS ...

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ChemQuest 15 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

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