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**Expected Learning Outcomes - Department of Chemistry - The ...**

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A. Buffer solution B. Neutral solution C. Saturated solution D. Supersaturated solution. 10. Protein is a polymer formed from the linkage of A. Amino acid molecules. B. Fatty acid molecules. C. Glucose units. D. Monosaccharides. **WAEC Chemistry Theory Questions . SECTION A. 1. Identify the solid remaining when each of the following is heated.**

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A solution of 0.20 mole of NaBr and 0.20 mole of MgBr 2 in 2.0 dm 3 of water is to be analysed. ... **WAEC 2020 Chemistry Theory Questions PAPER 2 (ESSAY) SECTION A. 1. Identify the solid remaining when each of the following is heated. ... State whether R would be expected to form acidic or basic oxide. (b) (i) State two assumptions of the ...**

**WAEC Chemistry Questions and Answers for 2020 (Theory and ...**

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An aqueous solution that contains 1 mol (342 g) of sucrose in enough water to give a final volume of 1.00 L has a sucrose concentration of 1.00 mol/L or 1.00 M. In chemical notation, square brackets around the name or formula of the solute represent the molar concentration of a solute. Therefore, [sucrose] = 1.00M.

**4.5: Concentration of Solutions - Chemistry LibreTexts**

A buffer solution (more precisely, pH buffer or hydrogen ion buffer) is an aqueous solution consisting of a mixture of a weak acid and its conjugate base, or vice versa.Its pH changes very little when a small amount of strong acid or base is added to it. Buffer solutions are used as a means of keeping pH at a nearly constant value in a wide variety of chemical applications.

**Buffer solution - Wikipedia**

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