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In physics, a partition function describes the statistical properties of a system in thermodynamic equilibrium. [citation needed] Partition functions are functions of the thermodynamic state variables, such as the temperature and volume. Most of the aggregate thermodynamic variables of the system, such as the total energy ...

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energy associated with temperature. Thermodynamics (from the Greek words meaning...

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Thermodynamics is a branch of physics that deals with heat, work, and temperature, and their relation to energy, radiation, and physical properties of matter. The behavior of these quantities is governed by the four laws of thermodynamics which convey a quantitative description using measurable macroscopic physical quantities, but may be explained in terms of microscopic constituents by ...

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Relation to heat and internal energy. In thermodynamics, heat is energy in transfer to or from a thermodynamic system, by mechanisms other than thermodynamic work or transfer of matter. Heat refers to a quantity transferred between systems, not to a property of any one system, or 'contained' within it. On the other hand, internal energy is a property of a single system.

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thermal physics followed by a presentation of the statistical theory, this book presupposes no exposure to statistics or quantum mechanics.

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