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want to carry out a

reaction that requires
combining one atom of

... Yes; the answer has

the correct units of

atoms of boron. Is the

number of significant

figures correct? Yes; the

mass of boron was

given to three

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1. Silver nitrate reacts with barium chloride to form silver chloride and barium nitrate. a. Write and balance the

chemical equation. 2

$\text{AgNO}_3 + \text{BaCl}_2 \rightarrow$

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$\text{AgCl} + \text{Ba}(\text{NO}_3)_2$ b. If
39.02 grams of barium
chloride are reacted in
an excess of silver
nitrate, how many

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dekaration 9.7 m/s 4.62

$\times 10^{-2}$ cm 6.75×10^{-4} g

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μg 7.8×10^3 s 2×10^2

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Solving continued

Sample Problem 1

Ammonia is made
industrially by reacting
nitrogen and hydrogen
under pressure, at high
temperature, and in the
presence of a catalyst.

The equation is $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$.

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