

## Holt Modern Chemistry Chapter 3

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History of the Alphabet: Part 3Chapter 5 2 Part 1 Chemistry SAT, part 1 of chp 7.3 Holt McDougal ~~OLD LIBRARY AMBIENCE: Rain Sounds, Book Sounds, Writing Sounds, Candle Flame Crackle~~ hydrogen, nitrogen,  $\text{He}$  helium emission spectra

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Holt Modern Chemistry Chapter 3 - Modularityscale

1. all matter is composed of atoms. 2. atoms of a given element are identical. 3. atoms can't be subdivided, created, or destroyed. 4. atoms of different elements combine in whole-number ratios to form chemical compounds. 5. atoms are combined, separated, or rearranged in chemical reactions. J. J. (Joseph John) Thomson.

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

3 Atoms: The Building Blocks of Matter

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CHAPTER 3 TEST Class Atoms: The Building Blocks of Matter MULTIPLE CHOICE On the line at the left of each statement, write the letter of the choice best completes the statement or answers the question. The behavior of cathode rays in a glass tube containing gas at low pressure led scientists to conclude that the rays were composed of a. energy

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