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Perform the following operations and express the
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106) $(.12 (5.35 \times 10^2) + (6.91 \times 10^{-2}) + (2.4 \times 10^{-3}))$

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MASTER Operations with Scientific Notation Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal. Not equal Equal $(3.4 \times 10^2) + (4.57 \times 10^3) = (0.34 \times 10^3) + (4.57 \times 10^3)$ The decimal is moved to the left to increase the exponent.

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Scientific notation takes the form of $M \times 10^n$ where $1 < M < 10$ and "n" represents the number of decimal places to be moved, Positive n Indicates the standard form Is a large number. Negative n Indicates a number between zero and one, Example 1: Convert 1,500,000 to scientific notation,

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SCIENTIFIC NOTATION Name Example 1: Convert 1,500,000 to ...

Math Handbook Transparency Masters Chemistry: Matter and Change 93 Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal. $(3.4 \times 10^2) + (4.57 \times 10^3) \neq (0.34 \times 10^3) + (4.57 \times 10^3)$ Not equal Equal The decimal is moved to the left to increase the exponent. $(0.34 \times 4.57) \times 10^3 = 4.91 \times 10^3$
Multiplication

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94 Chemistry: Matter and Change Math Handbook Transparency Worksheets 1. Perform the following operations and express the answers in scientific notation. a. $(1.2 \times 10^5) + 6(5.35 \times 10^4)$ $(1.2 \times 10^5) + (5.35 \times 10^6)$ $(0.12 \times 10^6) + (5.35 \times 10^6)$ $(0.12 \times 5.35) \times 10^6$ 5.47×10^6 b. $(6.91 \times 10^2) + (2.4 \times 10^3)$ $(6.91 \times 10^2) + (2.4 \times 10^3)$ $(6.91 \times 10^2) + (0.24 \times 10^2)$

Scientific Notation Use with Appendix B, Scientific

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Notation

96 Chemistry: Matter and Change Math Handbook
Transparency Worksheets 1. For each of the
measurements in the table below, determine if the
underlined number is significant or not significant.
Place a check mark in the appropriate box and in the
box under the rule that you used to make your
determination. 2.

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