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of mass &

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same ...

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of the Atom. Mass  
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Quantum Model.

Electron  
Configuration.

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The Building Blocks

of Matter. law of

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physical changes. e.g.  
20 g A + 20 g B 40 g  
AB.

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CHAPTER 3 REVIEW

Atoms: The Building  
Blocks of Matter

SECTION 2 SHORT

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ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

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3 Atoms: The Building  
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Atoms: The Building

Blocks of Matter law

of conservation of

massmass is neither

created nor destroyed

during ordinary

chemical reactions or

physical changes e.g.

20 g A + 20 g B 40 g

AB law of definite

proportions- a

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A chemical compound contains the same elements in exactly the same proportions by mass regardless of the size of the sample or the source of the compound e.g. If 10 grams of A combine with 20 grams of B to form compound AB, how ...

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Class Atoms: The Building Blocks of Matter MULTIPLE CHOICE On the line at the left of each statement, write the letter of the choice best completes the statement or answers the question. The

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behavior of cathode rays in a glass tube containing gas at low pressure led scientists to conclude that the rays were composed of a. energy

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