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1 of 10 The Periodic  
Table: Atomic Radius,  
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~~Chapter 7 – Periodic Properties of the~~

~~Elements: Part 1 of 11~~

~~Chapter 6 – The Electronic Structure of Atoms: Part 3 of 10~~

~~Chapter 6 – The Electronic Structure of Atoms: Part 8 of 10~~

Periodic Table

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Chapter 6 – The  
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Atoms: Part 9 of 10

Chapter 6 – The  
Electronic Structure of  
Atoms: Part 4 of 10

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STUDY. PLAY.

Electronegativity greater than 1.7 and is a transfer of electrons.

The Lower

Electronegativity

transfers to the higher electronegativity) t. Ionic Bond. Ions pack in a crystal lattice structure.

Lattice Energy.

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the space provided,

write the letter of the

term or phrase that best

completes each.

statement or best

answers each question.

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Chemistry 1. The charge on an ion is. a.always positive. b.always negative. c.either positive or negative.

Assessment Chapter  
Test A - Kettering City  
School District  
REVIEW Chemical  
Bonding MIXED  
REVIEW SHORT  
ANSWER Answer the  
following questions in

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the space provided. 1. a.

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6 Chemical Bonding -

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A, continued Use this

figure to answer

questions 7 and 8.

\_\_\_\_\_ 7. A solution

containing 35 g of Li

$2\text{SO}_4$  dissolved in 100

g of water is heated

from  $10^\circ\text{C}$  to  $90^\circ\text{C}$ .

According to

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information in the figure, this temperature change would result in

a. an additional 5 g of  $\text{Li}_2\text{SO}_4$  in ...

Assessment Chapter  
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16.1-16.3. White Board

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Name Section Quiz,

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continued Class Date 6.

Metals are malleable because when struck, one plane of metal atoms can slide past another plane without breaking bonds. b. cannot easily move out of the way. c. moves in a way that maximizes the repulsive forces within the

Home - David Brearley

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Multiple Choice Test ...

Chapter 12 Review 2

Multiple Choice Identify

the letter of the choice that best completes the statement or answers the question. 1. binds the atoms together is called a(n) A mutual electrical attraction between the nuclei and valence electrons of different atoms that a. dipole. c. chemical bond. b. Lewis

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