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~~3.2.1/3.2.2 Describe the differences between elements, compounds and mixtures.~~

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~~CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.~~

~~7 Chemical Formulas and Chemical Compounds~~

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CHAPTER 4 REVIEW Arrangement of Electrons in Atoms MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Under what conditions is a photon emitted from an atom? A photon is emitted when an electron moves from a higher energy level to a lower energy level. 2. What do quantum numbers describe?

4 Arrangement of Electrons in Atoms

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Modern Chemistry Chapter 9 Mixed Review Answers CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield.

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

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