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Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

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Assessment Chapter Test B - Ed W. Clark High School

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1.

Assessment Chapter Test A

Modern Chemistry Chapter Test B Answer Key Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution

Modern Chemistry Test B Answer Key Chapter10

Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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Modern Chemistry 104 Chapter Test Chapter: Solutions In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

_____ 1. Agitation prevents settling in a(n) a. alloy. b. homogeneous mixture. c. suspension. d. gaseous mixture. _____ 2. All of the following are heterogeneous mixtures except

Assessment Chapter Test A - Ed W. Clark High School

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