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Numbers • \_\_\_\_\_ • Used to \_\_\_\_\_ an  
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~~Note Taking Guide: Episode 304 Name  
Quantum Numbers~~

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Aditi Acharya Quantum Numbers  $n, l, m, s$   
Used to describe an electron in an atom  
 $n$  Principle quantum number Represents  
main energy level of electron maximum #

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of electrons in an Energy level =  $2n^2$

Example: What is the maximum number of electrons that can be in the 5th main energy level? 1 The 2 nd quantum number

- Describes the orbital shape within an energy level
- number of orbital shapes possible in energy level =  $n$  Orbital Shapes designated s,p ...

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~~note\_ep304aditi.pdf - orientation of orbital  
in space s ...~~

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Quantum Numbers  $n$   $l$   $m$   $s$  Used to  
describe an electron in an atom \_  $n$   
principle quantum \_ number Represents  
main energy level of \_ electron maximum

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# of \_electron \_ in an energy level  $= 2n^2$  \_ Example: What is the maximum number of electrons that can be in the 5th \_ main energy level? 1 The 2 nd quantum number \_\_\_\_\_ Describes the orbital \_\_\_\_\_ shape within an ...

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describe the location of the electron using

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and orbital notation. Aired:...

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