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Organic chemistry - Multiple choice questions - Sample ...
CHEMISTRY 2423 Practice FINAL EXAM A (Answers) PART

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- I. (2 points each) 1. (a) 4-bromo-3-ethylheptane (b) 1,3-dibromo-1,3-cyclohexadiene (c) trans-4-methyl-2-pentene
2. Br H C H 2 C 3 (a) C= CH 3 H | H 2 CH 3 (b) CH 3 - C - C
C - CH 2 - CH 3 CH 3 CH 3 CH 3 (c) PART II. (2 points each)
3. A 4. A 5. B 6. C 7. D 8. C 9. A 10. A 11. C 12. D 13. B 14.
B 15. B 16. A 17. B 18. D 19. D 20. A 21. D 22. D 23. D 24. D
25.

Sample Final Examination Organic Chemistry I

Three test tubes contain white crystalline organic solids A, B, and C, each of which melts at $149 - 150\text{ }^{\circ}\text{C}$ A 50 - 50 mixture of A and B melts at $130 - 139\text{ }^{\circ}\text{C}$ (In what range would a 5 ...

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Instructor: Michael C. Biewerbiewerm@utdallas.edu, BE 3.326. The following material can be downloaded as pdf files.

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June 2018 Exam / Mark Scheme / Model Answer. A-level Paper 2 ... Unit 4 - Rates, Equilibria and Further Organic Chemistry. Unit 5 - Transition Metals and Organic Nitrogen Chemistry Jan 2010 Exam / Mark Scheme June 2010 Exam / Mark Scheme Jan 2011 Exam / Mark Scheme

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Old Organic Chemistry Exams and Keys

Then you add the prefix for the # of carbons in the chain: 1 = meth 6 = hexa. 2 = eth 7 = hepta. 3 = prop 8 = octa. 4 = but 9 = nona or non. 5 = pent 10 = deca or dec. So if your molecule was...

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Organic chemistry - Six-mark questions. Six-mark questions are extended open response questions. These require longer answers than the structured questions with fewer marks.

Organic chemistry - Six-mark questions - Sample exam ...

Organic Chemistry 32-235 Practice Questions for Exam #2

Part 1: (Circle only ONE choice, circling more than one will be

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counted as wrong!) 4 points each 1. The correct IUPAC name for the following compound is: ... The answer is (C) because selectivity is high for bromination which prefers to attach the weaker C-H bond. 17.

Organic Chemistry 32-235 Practice Questions for Exam #2 ONE

1) Organic Chemistry is the study of Carbon containing compounds. 2) A covalent bond must be polar, 3) The Cahn-Ingold and Prelog convention describes rules for assigning (+) and (-) to chiral centers. 4) The Hammond Postulate states that entropy is inverted at a neutral chiral cation, 5) A typical C=C bond length is around 1.3×10^{-10} m.

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