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Trimethylamine is a pyramidal molecule.

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Solution: Dipole moment of CCl_4 is 0. The order of dipole moment is $\text{CCl}_4 < \text{CH}_3\text{Cl} < \text{CH}_2\text{Cl}_2 < \text{CHCl}_3$. Hence option (3) is the answer. 4. Which one of the following compounds will not be soluble in sodium bicarbonate ? (1) Benzene sulphonic acid (2) Benzoic acid (3) o-Nitrophenol (4) 2, 4, 6 - Trinitrophenol. Solution:

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