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Reviewing Key Concepts 1. (a) Sample answer: Bend the spoon, break it in pieces, and melt it. (b) Lighting a match is not a physical change. 2. (a) If a new substance is formed, a chemical change has occurred. (b) In electrolysis, water is broken down into hydrogen and oxygen, two different substances.

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Sample answer: An element is a pure substance that cannot be broken down into any other substances. A compound is a pure substance made of two or more elements chemically combined in a set ratio. 25.

Sample answer: According to the law of conservation of mass, matter is not created or destroyed in any chemical or physical change.

Chapter 1 Test – Introduction to Matter

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Observe changes in physical properties of materials. For instance, formation of a precipitate, gas production, and the color change are possible evidence that a chemical reaction has taken place.

CH 5 Lesson 1 (Observing Chemical Change) Flashcards | Quizlet

Recognizing Chemical Changes (page 54) 7. List the four possible clues to a chemical change? 8. Is the following statement true or false? If you observe a clue for chemical change, you can be certain that a chemical change has taken place. _____ 9. Define a precipitate. Conservation of Mass (page 55)

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SECTION 2.1 PROPERTIES OF MATTER (pages 39 – 42)

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2.4 Chemical Reactions and Enzymes
Lesson Objectives Explain how chemical

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Key Concepts
reactions affect chemical bonds. Describe how energy changes affect how easily a chemical reaction will occur. Explain why enzymes are important to living things.
Lesson Summary Chemical Reactions
Everything that happens in an organism is based on chemical reactions.

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Key: An equation that indicates how many molecules or atoms of each reactant and product are involved in a reaction. synthesis a chemical reaction in which two or more simple substances combine to form a new, more complex substance.

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