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Formulas 83 SECTION 9.3 NAMING
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MOLECULAR COMPOUNDS (pages
268–270) This section explains the
rules for naming and writing formulas
for binary

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The name of a base is the name of the cation followed by the name of the

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And Formulas Anions). The formula of a base is written by showing the number of hydroxide ions needed to balance the positive charge on the cation. After reading Lesson 9.4, answer the following questions.

Names and Formulas of Acids
hydrogen1.

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Chemical Names and Formulas - Mr.
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a.potassium sulfide e.hydrobromic
acidi.sulfur hexafluoride. b.tin(IV)
chloride f.aluminum
fluoride j.magnesium chloride.
c.hydrosulfuric acid g.dinitrogen

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pentoxide k. phosphoric acid. d. calcium
oxide h. iron(III) carbonate l. nitric acid.

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of figure 11.2 are Prentice Hall
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(9780132525770) Chapter 1
Introduction To Chemistry 89%
Complete. 1.1: The Scope of

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Pearson Chemistry Reading And Study
Workbook Answer Key

charges are anions. The names of
nonmetallic anions end in -ide.

Metallic cations take the name of the
metal. Some metals, including
transition metals, can form more than
one cation. Use a Roman number in

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the Stock name or use the classical name with a suffix to name these metals. 1

Chemical Names and Formulas -
Pittsfield

This is a list of the 118 chemical elements which have been identified

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as of 2020. A chemical element, often simply called an element, is a species of atoms which all have the same number of protons in their atomic nuclei (i.e., the same atomic number, or Z). A popular visualization of all 118 elements is the periodic table of the elements, a convenient tabular

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List of chemical elements - Wikipedia
PEARSON CHEMISTRY: The law of conservation of mass is explained in Lesson 2.4 (p. 50). Students are introduced to balancing chemical equations in Lesson 11.1 (pp. 346-

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354). Students learn to write and balance the five different types of chemical reactions in Lesson 11.2 (pp. 356-367). Students learn

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