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Answers will vary. There are two extra electrons in the ion compared to a neutral S atom. There are two more electrons than protons in the S²⁻ ion. Two electrons have been added to an atom of sulfur. 6. The dotted line around each polyatomic ion in Model 1 shows that the group of atoms has a charge.

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Acces PDF Polyatomic Ions Pogil Answer Key. group of atoms that are covalently bonded together and which carry an overall electrical charge. The ammonium ion, NH_4^+ , is formed when a hydrogen ion (H^+) attaches to the lone pair of an ammonia (NH_3) molecule in a coordinate covalent bond. Figure 1.

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O O. 1+. N H H H H H N S O O O. 1- 2- 1-. 2 POGIL™ Activities for High School Chemistry 2. Consider the terms “monatomic” and “polyatomic” as they are used in Model 1. Write a definition for each of these terms. It may be helpful to break the words apart (i.e., poly – atomic). Make sure your group comes to consensus.

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Chemical Bonding POGIL Activity 7 – A Mixed Up Activity Answer Key # 1 & #2. Chemical Formula Compound Name Monatomic Ions (#1) Polyatomic Ions (#2) CaSO₄ calcium sulfate calcium, Ca +2 sulfate, SO₄⁻² CaSO₃ calcium sulfite calcium, Ca +2 sulfite, SO₃⁻² Na₃PO₄ sodium phosphate sodium, Na +1 phosphate, PO₄⁻³ Li₂CO₃ lithium carbonate lithium, Li +1 carbonate, CO₃⁻² NH₄Cl ammonium chloride chloride, Cl -1 ammonium, NH₄⁺¹ Be(NO₂)₂ beryllium nitrite beryllium, Be +2 ...

~~POGIL_#7_answer_key—Chemical Bonding POGIL Activity 7 A ...~~

Pogil: Polyatomic Ion Key Based on your knowledge of monatomic ions, propose an explanation for the net charge on a polyatomic ion. Your answer should include a discussion of subatomic particles.

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