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hours. • Solute is the same substance in all beakers. Number of dissolved particles Beaker A Beaker B 1.0 g of solute added 2.0 g of solute added 5

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solution. Model 2 - Solute Dissolved vs. Solute Added The following data refer to an experiment in which a measured mass of solid is added to 10.0 q of 20 °C water.

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Unsaturated Solutions And Solubility ... Add a small amount of solute of known mass, if it dissolves it is unsaturated. If solid settles to the bottom filter the solution and mass the Page 20/47

remaining solid, if the mass is the same then the solution was saturated, if the mass is lower than it is an unsaturated solution. Model 2 - Solute Dissolved vs. Solute Added

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Model 1 Page 25/47

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Unsaturated Solutions Beaker A 1.0 g of solute added Beaker B 2.0 q of solute added Number of dissolved particles 5 Number of solid particles 0 Saturated Solutions Beaker CPage 27/47

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Answers Given scenarios, graphs, diagrams, or illustrations, the student will determine the type of solution such as saturated, supersaturated, or unsaturated.

Types of Page 30/47

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remaining solid if the mass is the same then the solution was saturated if the mass is lower than it is an unsaturated solution Model 2 - Solute Dissolved vs Solute Added

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