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TAGALOG G7

LECTURE #5:
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*Solutions,
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~~MODULE 4~~

~~ANSWER KEY |~~

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**Solubility in
different types of
solutions**

Saturated

Solutions Types of

solutions-

saturated,

unsaturated,

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*Supersaturated
Solution*

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~~SOLUTIONS~~

~~(WORKSHEET 5)~~

Unsaturated

Definition and

Example

Saturated and

unsaturated

solutions *Saturated*

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Model 1 -
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Unsaturated
beakers contain

10.0 g of water. •

All beakers are

kept at 20 °C. • All

solutions are

stirred for 2 hours.

- Solute is the same substance in all beakers.

Number of

dissolved particles

Beaker A Beaker B

1.0 g of solute

added 2.0 g of

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Solutions Pogil Saturated Answer Key File And Unsaturated Solutions

[klzz77r91vlg]

If solid settles to the bottom filter the solution and mass the remaining solid, if the mass is the same then the solution was

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unsaturated, if the mass is lower than it is an unsaturated solution. Model 2 –

Solute Dissolved vs. Solute Added

The following data refer to an experiment in which a measured mass of solid is added to 10.0 g of 20 °C water.

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Add a small amount of solute of known mass, if it dissolves it is unsaturated. If solid settles to the bottom filter the solution and mass the remaining solid, if the mass is

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the same then the
solution was
saturated, if the
mass is lower than
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solution. Model 2 -
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vs. Solute Added

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amount of solute
that will dissolve in
a solvent? We use

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a small amount of
solute of known

mass, if it dissolves
it is unsaturated If
solid settles to the
bottom filter the
solution and mass
the remaining

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Model 1 –

Saturated and
Unsaturated

- Solutions • All beakers contain 10.0 g of water. • All beakers are kept at 20 °C. • All solutions are stirred for 2 hours. • Solute is the same substance in all beakers.

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Solutions Beaker A

1.0 g of solute

added Beaker B 2.0

g of solute added

Number of

dissolved particles

5 Number of solid

particles 0

Saturated Solutions

Beaker C

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graphs, diagrams,
or illustrations, the
student will

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determine the type
of solution such as
saturated,
supersaturated, or
unsaturated.

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solution Model 2 -
Solute Dissolved vs
Solute Added

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