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Key Concepts 15.1 The high surface tension of water and low vapour pressure are due to the hydrogen bonding between the molecules The structure of ice is a regular open frame-work of water molecules held together by hydrogen bonds and arranged like a honeycomb Molecules in liquid water are packed closer together than they are in ice

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KOrriban. Prentice Hall Chemistry: Chapter 15. Surface Tension. Surfactant. Key Concepts 15.1. Aqueous Solution. The inward force that tends to minimize the surface area of a liquid. Any substance that interferes with the hydrogen bonding between water molecules. The high surface tension of water and low vapour pressure are examples.

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© 2009, Prentice-Hall, Inc. Chapter 15 Chemical Equilibrium John D. Bookstaver. St. Charles Community College. Cottleville, MO. Chemistry, The Central Science, 11th edition. Theodore L. Brown, H. Eugene LeMay, Jr., and Bruce E. Bursten

Chapter 15 Chemical Equilibrium

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This corrected second edition contains new material which includes solvent effects, the treatment of singlet diradicals, and the fundamentals of computational chemistry. "Computational Chemistry: Introduction to the Theory and Applications of Molecular and Quantum Mechanics" is an invaluable tool for teaching and researchers alike. The book provides an overview of the field, explains the basic underlying theory at a meaningful level that is not beyond beginners, and it gives numerous comparisons of different methods with one another and with experiment. The following concepts are illustrated and their possibilities and limitations are given: - potential energy surfaces; - simple and extended Hückel methods; - ab initio, AM1 and related semiempirical methods; - density functional theory (DFT). Topics are placed in a historical context, adding interest to them and removing much of their apparently arbitrary aspect. The large number of references, to all significant topics mentioned, should make this book useful not only to undergraduates but also to graduate students and academic and industrial researchers.

Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems. Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes. Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes. Chapters 6 to 8 look at acid-base equilibria and the link between electromagnetic radiation and the structure of atoms. Chapters 9 to 11 cover different types of bonding, the rates of chemical reactions, and the process of adsorption. Chapters 12 to 14 present molecular aggregates, magnetic resonance spectroscopy and photochemistry, and radiation. This book is useful to biological scientists for self-study and reference. With modest additions of mathematical material by the teacher, the book should also be suitable for a full-year major's course in physical chemistry.

Computational chemistry has become extremely important in the last decade, being widely used in academic and industrial research. Yet there have been few books designed to teach the subject to nonspecialists. Computational Chemistry: Introduction to the Theory and Applications of Molecular and Quantum Mechanics is an invaluable tool for teaching and researchers alike. The book provides an overview of the field, explains the basic underlying theory at a meaningful level that is not beyond beginners, and it gives numerous comparisons of different methods with one another and with experiment. The following concepts are illustrated and their possibilities and limitations are given: - potential energy surfaces; - simple and extended Hückel methods; - ab initio, AM1 and related semiempirical methods; - density functional theory (DFT). Topics are placed in a historical context, adding interest to them and removing much of their apparently arbitrary aspect. The large number of references, to all significant topics mentioned, should make this book useful not only to undergraduates but also to graduate students and academic and industrial researchers.

Authored by Paul Hewitt, the pioneer of the enormously successful "concepts before computation" approach, Conceptual Physics boosts student success by first building a solid conceptual understanding of physics. The Three Step Learning Approach makes physics accessible to today's students. Exploration - Ignite interest with meaningful examples and hands-on activities. Concept Development - Expand understanding with engaging narrative and visuals, multimedia presentations, and a wide range of concept-development questions and exercises. Application - Reinforce and apply key concepts with hands-on laboratory work, critical thinking, and problem solving.

Written for those less comfortable with science and mathematics, this text introduces the major chemical engineering topics for non-chemical engineers. With a focus on the practical rather than the theoretical, the reader will obtain a foundation in chemical engineering that can be applied directly to the workplace. By the end of this book, the user will be aware of the major considerations required to safely and efficiently design and operate a chemical processing facility. Simplified accounts of traditional chemical

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engineering topics are covered in the first two-thirds of the book, and include: materials and energy balances, heat and mass transport, fluid mechanics, reaction engineering, separation processes, process control and process equipment design. The latter part details modern topics, such as biochemical engineering and sustainable development, plus practical topics of safety and process economics, providing the reader with a complete guide. Case studies are included throughout, building a real-world connection. These case studies form a common thread throughout the book, motivating the reader and offering enhanced understanding. Further reading directs those wishing for a deeper appreciation of certain topics. This book is ideal for professionals working with chemical engineers, and decision makers in chemical engineering industries. It will also be suitable for chemical engineering courses where a simplified introductory text is desired.

Discusses the formation, composition, properties and processing of the principal fossil and biofuels, ideal for graduate students and professionals.

Presents aquatic chemistry in a way that is truly useful to those with diverse backgrounds in the sciences. Major improvements to this edition include a complete rewrite of the first three background chapters making them user-friendly. There is less emphasis on mathematics and concepts are illustrated with actual examples to facilitate understanding.

Prentice Hall Physical Science: Concepts in Action helps students make the important connection between the science they read and what they experience every day. Relevant content, lively explorations, and a wealth of hands-on activities take students' understanding of science beyond the page and into the world around them. Now includes even more technology, tools and activities to support differentiated instruction!

Organic Chemistry, A Modern Approach, Vol-II, is for the second year students pursuing BSc Chemistry (Honours) at the University of Calcutta and other major universities across eastern India. It offers "learning by practice" approach. Supplemented with 500+ reaction mechanisms with curved-arrow notation, the book lays a solid foundation for advanced aspects of the subject-matter.

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