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Electrons in
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*Pearson Chapter
Page 5/44*

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5: Section 1:

*Revisiting the
Atomic Model*

Zumdahl

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ed. Chapter 5

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~~5~~

~~Thermochemistry~~

~~(Sections 5.1 —~~

~~5.4) Chapter 5 —~~

~~Molecules and~~

~~Compounds Ch 5~~

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~~History of the
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Pearson Chapter
5: Section 3:~~

Atomic Emission
Spectra and the
Quantum

Mechanical Model

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~~Chapter 5:~~

~~Periodic Law~~

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~~minutes or less)~~

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Chapter (5)

Thermochemical

Equations

Practice

Problems ~~The~~

~~Photoelectric~~

~~Effect~~ ~~?????~~ 5

~~??????~~ Zumdahl

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~~The Scope of~~

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~~Electrons in~~

~~Atoms Pt II~~

~~Chapter 5~~

~~Thermochemistry~~

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~~(Sections 5.5~~

~~5.7)~~ **Zumdahl**
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Arrangements in

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-

Thermochemistry:

Part 1 of 8 *CHEM*

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terms or answers
are repeated
throughout—in
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questions)

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energy levels.

Fixed energies
an electron can
have are called?

no. Can

electrons in an
atom be between
energy levels?

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questions) Terms
in this set (53)
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an electron can
have are called?
no.

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Ground State
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atomic orbital.
a mathematical
expression
describing the
probability of
finding an

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Chemistry at
various
locations;
usually

represented by
the region of
space around the
nucleus where
there is a high
probability of
finding an
electron.

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