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in Advances in Quantum Chemistry, edited by P.-O. Löwdin, volume 34, pages 143-269. Academic Press, New York, 1999. Academic Press, New York, 1999. David Sherrill 2006-08-15

The biggest change in the years since the first edition is the proliferation of computational chemistry programs that calculate molecular properties. McQuarrie presents step-by-step SCF calculations of a helium atom and a hydrogen molecule, in addition to including the Hartree-Fock method and post-Hartree-Fock methods.

Undergraduate level text including problems and answers.

Praised for its appealing writing style and clear pedagogy, Lowe's Quantum Chemistry is now available in its Second Edition as a text for senior undergraduate- and graduate-level chemistry students. The book assumes little mathematical or physical sophistication and emphasizes an understanding of the techniques and results of quantum chemistry, thus enabling students to comprehend much of the current chemical literature in which quantum chemical methods or concepts are used as tools. The book begins with a six-chapter introduction of standard one-dimensional systems, the hydrogen atom, many-electron atoms, and principles of quantum mechanics. It then provides thorough treatments of variation and perturbation methods, group theory, ab initio theory, Huckel and extended Huckel methods, qualitative MO theory, and MO theory of periodic systems. Chapters are completed with exercises to facilitate self-study. Solutions to selected exercises are included. Assumes little mathematical or physical sophistication Emphasizes understanding of the techniques and results of quantum chemistry Includes improved coverage of time-dependent phenomena, term symbols, and molecular rotation and vibration Provides a new chapter on molecular orbital theory of periodic systems Features new exercise sets with solutions Includes a helpful new appendix that compiles angular momentum rules from operator algebra

The detailed solutions manual accompanies the second edition of McQuarrie's Quantum Chemistry.

Advanced graduate-level text looks at symmetry, rotations, and angular momentum addition; occupation number representations; and scattering theory. Uses concepts to develop basic theories of chemical reaction rates. Problems and answers.

Introduction to problems of molecular structure and motion covers calculus of orthogonal functions, algebra of vector spaces, and Lagrangian and Hamiltonian formulation of classical mechanics. Answers to problems. 1966 edition.

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This text provides students with concise reviews of mathematical topics that are used throughout physical chemistry. By reading these reviews before the mathematics is applied to physical chemical problems, a student will be able to spend less time worrying about the math and more time learning the physical chemistry.

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