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(A) less than 33. (A) Organic chemistry
34. (C) mineral 35. (D) Air. GENERAL
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... Questions and Answers 1. The
particle that has the smallest mass is
the. A. ... The concept of multiple

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Covalent bonds is used to explain the molecular formula of . A. O₂. B. NaCl. C. C₂H₆. D. F₂. E.

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Choice Questions Question 1 Which of the following terms are unitless? (a) Molality (b) Molarity ... How many mL of water must be added to 200 mL of .65 M HCL to dilute the solution to .20 M (a) 450 mL (b) 400 mL (c) 350 mL (d) 500 mL

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Multiple Choice Questions On
Chemical bonding; Molarity -
Concentration of Solutions; Dilution of
Solutions; Avogadro's Number and
the Molar Mass of an Element; The
Structure of Ethene (Ethylene): sp^2

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Hybridization; Rate of radioactive decay and calculation of Half-life time; Atomic Number, Mass Number, and Isotopes; MCQ on Chapter: Thermochemistry

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NCERT Class 12 Chemistry Multiple
Choice Questions(MCQs) & Answers

NCERT Class 12 Chemistry

QUESTIONS NCERT Class 12

Chemistry Multiple Choice

Questions(MCQs) & Answers 21

Trunbull's blue is A Ferrous

ferricyanide B $\text{Fe}_3[\text{Fe}(\text{CN})_6]_4$ C

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Questions
Ferrous cyanide D Ferricyanide

Answer: Ferrous ferricyanide 22

Roasted copper pyrites on smelting with sand produces A FeSiO_3 and Cu_2S B $\text{Fe}_3(\text{PO}_4)_2$...

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