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$0.86163 \cos(31.5) = 0.85264 \cos(31) = 0.86$ 15. Some elementary chemistry textbooks give the value of R, the ideal gas constant, as 0.0821 l atm K⁻¹ mol⁻¹.

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Solutions to exercises 1B.1(b) The mean speed is $8RT/v_{\text{mean}} M$ The mean translational kinetic energy is $m \cdot 3RT/2$ $3kT/2$ $M/2$ The ratios of species 1 to species 2 at the same temperature are $E_k/v_{\text{mean},1}$ $v_{\text{mean},2}$ (i) $1/2$ $m v_2^2/M_1$ $v_{\text{mean},H}$ $1/2$ $m v_2^2$ and $2/v_{\text{mean},Hg}$ $200.6/4.003$ $1/2$ $2/m v_{\text{rms}}$ $E_k/1$ $E_k/2$ 7.079 (ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

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