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Answer: 4.5oC. Measuring Heat of Reaction for a simple system. An important thermodynamic property in a reaction is the molar enthalpy of reaction (q_{sys}). The molar enthalpy of reaction is defined as the heat absorbed by the system from the surroundings in a reaction when one mole of product is formed.

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General Chemistry: Pearls of Wisdom is a review manual that contains valuable help for students preparing for the AP Chemistry Test, MCAT, VCAT and DCAT, as well as improving their chemistry knowledge base. Written in a rapid-fire question and answer format, this study aid will provide you with the instant gratification of a correct answer. This book emphasizes key facts that may be overlooked, quickly forgotten and that somehow seem to be needed on the MCAT, VCAT, DCAT and similar examinations.

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This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid/base reactions which often persists through working with acid/base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshman in STEM fields.

More people get into medical school with a Kaplan MCAT course than all major courses combined. Now the same results are available with MCAT General Chemistry Review. This book features thorough subject review, more questions than any competitor, and the highest-yield questions available. The commentary and instruction come directly from Kaplan MCAT experts and include targeted focus on the most-tested concepts. MCAT General Chemistry Review offers: UNPARALLELED MCAT

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KNOWLEDGE: The Kaplan MCAT team has spent years studying every MCAT-related document available. In conjunction with our expert psychometricians, the Kaplan team is able to ensure the accuracy and realism of our practice materials. **THOROUGH SUBJECT REVIEW:** Written by top-rated, award-winning Kaplan instructors, all material has been vetted by editors with advanced science degrees and by a medical doctor. **EXPANDED CONTENT THROUGHOUT:** As the MCAT has continued to develop, this book has been updated continuously to match the AAMC's guidelines precisely—no more worrying if your prep is comprehensive! **STAR RATINGS FOR EVERY SUBJECT:** New for the 3rd Edition of MCAT General Chemistry Review, every topic in every chapter is assigned a star rating—informed by Kaplan's decades of MCAT experience and facts straight from the testmaker—of how important it will be to your score on the real exam. **MORE PRACTICE THAN THE COMPETITION:** With 350+ questions throughout the book and access to a full-length practice test online, MCAT General Chemistry Review has more practice than any other MCAT general chemistry book on the market. **ONLINE COMPANION:** One practice test and additional online resources help augment content studying. The MCAT is a computer-based test, so practicing in the same format as Test Day is key. **TOP-QUALITY IMAGES:** With full-color, 3-D illustrations, charts, graphs and diagrams from the pages of Scientific American, MCAT General Chemistry Review turns even the most intangible, complex science into easy-to-visualize concepts. **KAPLAN'S MCAT REPUTATION:** Kaplan is a leader in the MCAT prep market, and twice as many doctors prepared for the MCAT with Kaplan than with any other course.* **UTILITY:** Can be used alone or with the other companion books in Kaplan's MCAT Review series. * Doctors refers to US MDs who were licensed between 2001-2010 and used a fee-based course to prepare for the MCAT. The AlphaDetail, Inc. online study for Kaplan was conducted between Nov. 10 - Dec. 9, 2010 among 763 US licensed MDs, of whom 462 took the MCAT

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The MCAT is a test of more than just the facts about basic physical and biological sciences—it's an in-depth, rigorous examination of your knowledge of scientific concepts and principles, as well as your critical-thinking and writing skills. With the Princeton Review's subject-specific MCAT series, you can focus your review on the MCAT topics that are most challenging to you. Each book in the series contains the most in-depth coverage of subjects tested on the MCAT. Each chapter in MCAT General Chemistry includes:

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MCAT General Chemistry Review also includes:

- A complete glossary of general chemistry terms
- A general chemistry formula sheet

Interactive General Chemistry meets students where they are...with a general chemistry program designed for the way students learn. Achieve provides a new platform for Interactive General Chemistry, thoughtfully developed to engage students for better outcomes. Powerful data and analytics provide instructors with actionable insights on a platform that allows flexibility to align with a broad variety of teaching and learning styles and the exciting Interactive General Chemistry program! Whether a student's learning path starts with problem solving or with reading, Interactive General Chemistry delivers the learning experience he or she needs to succeed in general chemistry. Built from the ground up as a digital learning program, Interactive General Chemistry combines the Sapling Learning homework platform with a robust e-book with seamlessly embedded, multimedia-rich learning

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resources. This flexible learning environment helps students effectively and efficiently tackle chemistry concepts and problem solving. Student-centered development In addition to Macmillan's standard rigorous peer review process, student involvement was critical to the development and design of Interactive General Chemistry. Using extensive research on student study behavior and data collection on the resources and tools that most effectively promote understanding, we crafted this complete course solution to intentionally embrace the way that students learn. Digital-first experience Interactive General Chemistry was built from the ground up to take full advantage of the digital learning environment. High-quality multimedia resources--including Sapling interactives, PhET simulations, and new whiteboard videos by Tyler DeWitt--are seamlessly integrated into a streamlined, uncluttered e-book. Embedded links provide easy and efficient navigation, enabling students to link to review material and definitions as needed. Problems drive purposeful study Our research into students' study behavior showed that students learn best by doing--so with Interactive General Chemistry, homework problems are designed to be a front door for learning. Expanding upon the acclaimed Sapling homework--where every problem contains hints, targeted feedback, and detailed step-by-step solutions--embedded resources link problems directly to the multimedia-rich e-book, providing just-in-time support at the section and chapter level.

The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success, Volume 1 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses. The book provides easy-to-understand, step-by-step instructions for solving general chemistry problems. The book begins with chapters dedicated to problem solving methodology and unit conversions. In subsequent chapters, the text covers important topics like ionic and covalent bonding, chemical formula calculations, solubility

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and reactions in aqueous solution, gases, the first law of Thermodynamics, Quantum theory, and electron configuration. It also covers periodic trends, the Lewis Dot Structures, and bonding theories. Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand. The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry. Rachel Turoscy earned her B.S., M.S., and Ph.D. in chemistry from Lehigh University. She is an associate professor of chemistry at Middlesex County College. Dr. Turoscy has authored numerous articles, book chapters, and textbooks addressing various topics in the world of chemistry.

This print companion to MindTap General Chemistry: Atoms First presents the narrative, figures, tables and example problems—but no graded problems or assessments. Students must use MindTap to complete the interactive activities, exercises, and assignments. The atoms first organization introduces students to atoms and molecules earlier and delays math-intensive problem-solving to later in the semester. This gives students a stronger conceptual framework to help them succeed in the course. In addition, the narrative provides greater emphasis on the historical development of the atomic nature of matter and atomic structure. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

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