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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals. $\text{Br (l)} + 2\text{NaI(s)} \rightarrow 2\text{NaBr(s)} + \text{I}_2\text{(s)}$ 3. An aluminum strip is immersed in a solution of silver nitrate.

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The Ultimate Chemical Equations Handbook SnCO tin(II) carbonate sodium hydrogen carbonate NaHCO 1. 2. 3. 4. 5. 6. 8. 9. 10. 11. vanadium(V) oxide H₂O dihydrogen monoxide (NH) CO ammonium oxalate 4 224 polonium(VI) thiocyanate Po(SCN)₆ tetraphosphorus decaoxide P O 4 10 12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

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The Ultimate Chemical Equations Handbook Student Edition By George R Hague, Jr and Jane D Smith question requiring the determination of products of chemical reactions (the Equations section) five of these eight equations to answer (Note: Additional responses will not be graded or counted If all [PDF] Ultimate Chemical Equations Handbook Answer Key

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